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chapter 1 (Fundamental Concepts)
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- 1- every sample has exact same characteristic physical and chemical properties used to identify substance
- 2- every sample of a pure substance has exact same composition with the same makeup (i.e.: pure water is

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always 11.2% H and 88.8% oxygen by mass

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This topic covers 1.2 Bonding and Structure and 1.3.1 Periodicity (melting and boiling trends only) Bonding,

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Chemistry Chapter 1 Test Review
Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct answer in the blank. _____ 1. Inorganic chemistry is the study of a. non-carbon related compounds. b. the chemistry of living

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things. c. mathematical modeling. d.

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Molecular Weight, Mass Percentage of
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2 ANSWERS TO EXAM-STYLE
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UNIVERSITY PRESS 2011 molar mass of $I_2 = 253.80 \text{ g mol}^{-1}$ mass of $I_2 = 1.25 \times 10^{-3} \times 253.80$, i.e. 0.317 g
16 a molar mass of $PbI_2 = 460.99$
 $0.1270 \text{ moles of } PbI_2 = 460.99 = 2.755 \times 10^{-4} \text{ mol}$
b $Pb(NO_3)_2(aq) + MI_2(aq) \rightarrow PbI_2(s) + M(NO_3)_2(aq)$

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